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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.006 \AA$
$R$ factor $=0.070$
$w R$ factor $=0.245$
Data-to-parameter ratio $=12.6$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## 5,6-Dihydrobenzo[h]quinazoline

5,6-Dihydrobenzo[ $h$ ]quinazoline, $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2}$, has been synthesized from $\alpha$-tetralone and formamide using different palladium complexes as catalysts. There are two molecules in the asymmetric unit and the least-squares planes through the two molecules show that they are almost perpendicular to each other, resulting in a zigzag packing pattern in the crystal structure.

## Comment

Pyrimidines are widely found in nature, e.g. in pyrimidine bases and purines in nucleic acids, and in the vitamin thiamin. They have also attracted interest as potential drugs and are available in the nucleoside analogue AZT used in AIDS therapy, Acyclovir used in treatment of herpes infections and the prodrug Capecitabine used in cancer therapy. The interesting chemical and physiological properties of pyrimidines have led to a number of syntheses being developed [see von Angerer (2004) for a comprehensive review]. In our procedure, good yields of the expected products are formed from the appropriate ketone when reacted with formamide, catalyzed by different palladium complexes (Ingebrigtsen et al., 2005). The title compound, (I), crystallizes in the monoclinic centrosymmetric space group $P 2_{1} / c$, with two molecules in the asymmetric unit. The atomic numbering scheme of the title compound is shown in Fig. 1. The bond lengths are within the normal range of such bonds (Allen et al., 1987). There are no significant $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ or $\mathrm{C}-\mathrm{H} \cdots \pi($ ring $)$ interactions between the molecules in the crystal packing. The benzene and pyrazine rings in each independent molecule are essentially planar, while they make a dihedral angle with each other of 17.5 (2) and 19.4 (2) for molecules 1 (N11) and 2 (N21), respectively. The total puckering amplitude parameter $Q_{\mathrm{T}}$ for the central six-membered ring system is 0.470 (3) and 0.468 (3) $\AA$, respectively, for molecule 1 and 2 (Cremer \& Pople, 1975; Iulek \& Zukerman-Schpector, 1997). This ring puckering is described as $50 \%$ half-chair and $44 \%$ twisted-boat for molecule 1, and $47 \%$ half-chair and $45 \%$ twisted-boat for molecule 2. Fig. 2 shows the zigzag packing of the molecules in the crystal structure.

(I)

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## Experimental

To a 10 ml flask charged with $\mathrm{Pd}(\mathrm{OAc})_{2}\left(0.05\right.$ equivalents) and $\mathrm{PPh}_{3}$ ( 0.10 equivalents) were added formamide $(5.0 \mathrm{~g}), \mathrm{PhI}(2.0 \mathrm{~g})$ and $\alpha$ tetralone ( 1.0 equivalents). The resulting mixture was heated at 433 K for 8 h . The reaction mixture was diluted with diethyl ether and extracted three times with 2 MHCl . The combined aqueous layers were basified with $4 M \mathrm{NaOH}$ and extracted with diethyl ether. The organic layer was washed with water and brine, and dried over $\mathrm{Na}_{2} \mathrm{CO}_{3}$. Evaporation of the solvent gave the crude product as a white solid. Purification by silica-gel column chromatography (EtOAc) gave crystals that were dissolved in a small amount of diethyl ether. Heptane was added and crystals of the title compound were grown by slow evaporation of the solvent at room temperature.

## Crystal data

$\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2}$
$M_{r}=182.22$
Monoclinic, $P 2_{1} / c$
$a=25.118$ (5) $\AA$
$b=7.3522$ (9) $\AA$
$c=10.0830$ (17) $\AA$
$\beta=101.664$ (11) ${ }^{\circ}$
$V=1823.6(5) \AA^{3}$
$Z=8$

## Data collection

Enraf-Nonius CAD-4
diffractometer
$\omega-2 \theta$ scans
Absorption correction: $\psi$ scan
[ABSCALC in OSCAIL;
McArdle \& Daly (1999) and
North et al. (1968)]
$T_{\text {min }}=0.961, T_{\text {max }}=0.976$
3334 measured reflections
3184 independent reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.070$
$w R\left(F^{2}\right)=0.245$
$S=1.07$
3184 reflections
253 parameters
H -atom parameters constrained
$D_{x}=1.327 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 25 reflections
$\theta=12-16^{\circ}$
$\mu=0.08 \mathrm{~mm}^{-1}$
$T=298$ (2) K
Block, brown
$0.5 \times 0.3 \times 0.3 \mathrm{~mm}$

1703 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.006$
$\theta_{\text {max }}=25.0^{\circ}$
$h=-12 \rightarrow 29$
$k=0 \rightarrow 8$
$l=-11 \rightarrow 11$
3 standard reflections frequency: 120 min intensity decay: $10 \%$

$$
\begin{aligned}
& w= 1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.1448 P)^{2}\right. \\
&+0.3821 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.004 \\
& \Delta \rho_{\max }=0.44 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.22 \mathrm{e}^{-3}
\end{aligned}
$$

All the H atoms were positioned geometrically and refined using a riding model, fixing the aromatic $\mathrm{C}-\mathrm{H}$ distances at $0.93 \AA$ and the $\mathrm{CH}_{2} \mathrm{C}-\mathrm{H}$ distances at $0.97 \AA$, with $U_{\text {iso }}(\mathrm{H})=1.3 U_{\text {eq }}(\mathrm{C})$. The quality of the crystal was rather poor and accordingly data were collected only to $\theta_{\max }=25.0^{\circ}$.

Data collection: CAD-4-PC Software (Enraf-Nonius, 1992); cell refinement: CELDIM in CAD-4-PC Software; data reduction: XCAD4 (McArdle \& Higgins, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEX (McArdle, 1993a); software used to prepare material for publication: OSCAIL (McArdle, 1993b).


Figure 1
A view of the asymmetric unit of the title compound, (I), with the atomic numbering scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level.


A view of the crystal packing in the title compound, (I). Displacement ellipsoids are drawn at the $30 \%$ probability level.

## References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. \& Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.

Angerer, S. von (2004). Methods Synth. 16, 379-572.
Cremer, D. \& Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1359.
Enraf-Nonius. (1992). CAD-4-PC Software. Version 1.1. Enraf-Nonius, Delft, The Netherlands.
Ingebrigtsen, T., Helland, I. \& Lejon, T. (2005). Heterocyles, 65. Accepted.
Iulek, J. \& Zukerman-Schpector, J. (1997). Quim. Nova, 20, 433-434.
McArdle, P. (1993a). J. Appl. Cryst. 26, 752.
McArdle, P. (1993b). OSCAIL. Version 9. NUI Galway, Ireland.
McArdle, P. \& Daly, P. (1999). ABSCALC. PC version. NUI Galway, Ireland.
McArdle, P. \& Higgins, T. (1995). XCAD4. NUI Galway, Ireland.
North, A. C. T., Phillips, D. C. \& Mathews, F. S. (1968). Acta Cryst. A24, 351359.

Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.

